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²H-DNP-enhanced ²H-¹³C solid-state NMR correlation spectroscopy

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Perdeuteration of biological macromolecules for magic angle spinning solid-state NMR spectroscopy can yield high-resolution ${}^{2}H^{-13}C$ correlation spectra and the method is therefore of great interest for the structural biology community. Here we demonstrate that the combination of sample deuteration and dynamic nuclear polarization yields resolved ${}^{2}H^{-13}C$ correlation spectra with a signal enhancement of $\varepsilon \ge 700$ compared to a spectrum recorded with microwaves off and otherwise identical conditions. To our knowledge, this is the first time that ${}^{2}H$ -DNP has been employed to enhance MAS-NMR spectra of a biologically relevant system. The DNP process is studied using several polarizing agents and the technique is applied to obtain ${}^{2}H^{-13}C$ correlation spectra of U-[${}^{2}H$, ${}^{13}C$] proline.

Introduction

In recent years magic angle spinning NMR (MAS-NMR) spectroscopy has emerged as a valuable method to determine atomic-resolution structures of biomolecular macromolecules such as globular proteins, membrane proteins and amyloid fibrils.^{1,2} However, in contrast to solution-state NMR, the majority of MAS-NMR experiments rely on recording homoand heteronuclear ¹³C and ¹⁵N correlation spectra because direct ¹H detection is often compromised by the strong ¹H⁻¹H dipolar interactions present in the solid state. Under typical experimental conditions, these strong couplings result in broad, unresolved ¹H resonances. Techniques such as ultrafast sample spinning,^{3,4} windowed homonuclear decoupling techniques, 5,6 and dilution of the ${}^{1}H{-}^{1}H$ dipolar bath by deuteration can be used to narrow ¹H lines in MAS-NMR experiments and are currently under investigation.7-10 Successful implementation of these techniques would bring the resolving power of a third nucleus to MAS-NMR protein investigations.

Another approach to access a third nucleus is to observe deuterons (²H) because of their reduced homonuclear dipolar coupling that can be attenuated under moderate MAS frequencies (\sim 5 kHz). Deuterons contain similar information on the chemical environment as protons, and can therefore be directly employed to obtain structural information.

Recently it was shown that spectra of deuterated proteins exhibit high-resolution MAS-NMR spectra and the method is therefore of great interest for the structural biology community.^{8,11} Furthermore, deuteration can also result in additional benefits in both the resolution and sensitivity of more conventional ¹³C and ¹⁵N MAS-NMR experiments. For example, the resolution of 3D or 4D ¹³C spectra of deuterated proteins is no longer limited by the ¹H decoupling power and resulting rf heating. In addition, cross-polarization (CP) enhancements are increased and neither ¹H nor ¹³C longitudinal relaxation times are significantly increased.¹² However, the ²H quadrupole coupling ($e^2qQ/h \sim 167$ kHz for a CD bond) often reduces the sensitivity and resolution of directly observed ²H spectra in solids. At the same time, the relaxation and lineshape properties of the deuterium nucleus are particularly sensitive to the local dynamics and can provide valuable information.⁸

To overcome the difficulties associated with the deuterium quadrupole coupling, techniques such as rotor-synchronized pulse sequences or indirect detection through for example ¹³C can be used. Furthermore, in heteronuclear correlation experiments (*e.g.* ²H–¹³C), MAS narrows the first order ²H quadrupole interaction and the resolution can be further improved if a ²H double-quantum (²H-DQ) excitation and reconversion scheme is employed. ^{11,13,14}

NMR signal intensities of solids and liquids can be enhanced by several orders of magnitude with dynamic nuclear polarization (DNP)^{15,16} and in the last decade high-frequency DNP has emerged as a valuable method for a variety of applications, spanning particle physics,^{17,18} pharmaceutical applications^{19,20} and structural and mechanistic studies of biologically relevant molecules.^{15,21,22}

In a DNP experiment, the large thermal polarization of a paramagnetic polarizing agent is transferred to surrounding nuclei by microwave irradiation of the sample at the electron paramagnetic resonance (EPR) transition. DNP enhancements are measured by taking the ratio of signal intensity in spectra with and without microwaves, leaving all other experimental parameters unchanged. Depending on the inhomogeneous breadth of the EPR spectrum (Δ) and the homogeneous linewidth (δ), DNP can either occur through the solid-effect (SE) if the nuclear Larmor frequency ω_{0I} is larger than the EPR linewidth ($\omega_{0I} > \Delta, \delta$), or through the much more efficient cross-effect (CE) if $\Delta > \omega_{0I} > \delta$.^{15,25} In the classical description of the CE the underlying mechanism is a two-step process involving two electrons with Larmor frequencies ω_{0S1} and ω_{0S2} , and a nucleus with a frequency ω_{0I} . Initially, the allowed EPR transition of one electron is irradiated and nuclear

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polarization is generated in a subsequent three-spin flip-flop process through transitions such as $|\alpha_{1S}\beta_{2S}\beta_I\rangle \leftrightarrow |\beta_{1S}\alpha_{2S}\alpha_I\rangle$ or $|\beta_{1S}\alpha_{2S}\beta_I\rangle \leftrightarrow |\alpha_{1S}\beta_{2S}\alpha_I\rangle$.^{26,27} The maximum DNP enhancement is achieved when the difference between the electron Larmor frequencies of two electron spin packets satisfy the matching condition $|\omega_{0S1} - \omega_{0S2}| = \omega_{0I}$, with ω_{0I} the nuclear Larmor frequency. The DNP-enhanced nuclear polarization then disperses throughout the bulk *via* spin-diffusion.²⁸ Currently, the largest signal enhancements in solids at high magnetic fields (> 5 T) are observed in experiments where the cross-effect (CE) is the dominant DNP mechanism.^{23,24}

Here we demonstrate that the combination of sample deuteration and DNP yields resolved ${}^{2}H^{-13}C$ correlation spectra with a signal enhancement of $\varepsilon \ge 700$. To our knowledge, ${}^{2}H$ -DNP has been reported only for the preparation of polarized targets^{29–31} and in dissolution DNP,³² focusing on the polarization of small alcohol molecules. In this study, we demonstrate that high-field ${}^{2}H$ -DNP can be used to enhance MAS-NMR spectra of biologically relevant molecules. Although the technique is initially demonstrated using a single amino acid residue, the concept has considerable potential for structural investigations of biologically relevant macromolecules in the solid state at high magnetic fields. Given sufficient sensitivity, the resolving power of ${}^{2}H$, ${}^{13}C$ and ${}^{15}N$, 3D and 4D experiments have the potential to extend MAS-NMR to the application of larger biological systems.

Results and discussion

Polarizing agents and DNP-enhancement profiles

The molecular structures of the two polarizing agents TOTAPOL (1-(TEMPO-4-oxy)-3-(TEMPO-4-amino)-propan-2-ol) and OX063 (methyl-tris[8-carboxy-2,2,6,6-tetrakis[(2-hydroxyethyl]-benzo-[1,2-d:4,5-d]bis[1,3]dithiol-4-yl]) are shown in Fig. 1 and both are soluble in aqueous media at high concentration.

The 140 GHz (5 T) EPR spectra of TOTAPOL and OX063 are shown in Fig. 2 (top). While the EPR spectrum of TOTAPOL shows a large g-anisotropy and additional features due to the ¹⁴N hyperfine interaction with the electron spin,³³ the EPR spectrum of OX063 appears almost symmetric at high magnetic fields because no significant hyperfine couplings are present and the g-tensor anisotropy is small.³⁴ With an inhomogeneous breadth of $\Delta \approx 600$ MHz and 55 MHz for TOTAPOL and OX063, respectively, and a ²H nuclear Larmor frequency at 5 T of 32 MHz, we see that both radicals satisfy the conditions ($\Delta > \omega_{0I} > \delta$) for CE DNP for ²H.

The field-swept DNP enhancement profile is closely related to the high-field EPR spectrum recorded at the same magnetic field strengths as shown in Fig. 2. Typically high-field DNP experiments are performed using a fixed-frequency microwave



Fig. 1 Molecular structures of the two polarizing agents TOTAPOL and OX063.



Fig. 2 Top: Two-pulse echo-detected 140 GHz EPR spectra of 1 mM TOTAPOL and OX063 in glycerol/H₂O (60/40), T = 20 K. Bottom: Direct detected ²H-DNP enhancement profiles of 20 mM TOTAPOL and 40 mM trityl (OX063) in d₈-glycerol/D₂O (60/40) using a rotor-synchronized quadrupole-echo sequence. T = 90 K, $t_p(\pi/2) = 3$ µs, $\tau = 166$ µs, $\omega_r/2\pi = 6$ kHz. For comparison the DNP enhancement profiles are normalized to maximum intensity.

source and the DNP process needs to be optimized with respect to the magnetic field to find the best irradiation frequency.

In addition to determining the optimum field position for DNP, the enhancement profile also reveals much information about the nature of the underlying DNP process. Since both enhancement profiles of TOTAPOL and OX063 do not show resolved features at frequencies corresponding to $\omega_{0S} \pm \omega_{0I}$, it can be directly concluded that the underlying DNP mechanism observed in experiments reported here is the CE.^{18,25,27,35–37}

The DNP enhancement profile for TOTAPOL resembles the shape typically observed for TEMPO based (bi)radicals.38-40 For ²H-DNP the maximum negative enhancement is obtained at the low-field side of the profile corresponding to 4968.6 mT (DNP(-)), while the maximum positive enhancement is observed at 4979.1 mT (DNP(+)). This is in contrast to ¹H-DNP, where the overall maximum ¹H enhancement is observed at the high-field side (DNP(+)) of the DNP enhancement profile.^{38,41} Note that the ²H-DNP enhancement profile for TOTAPOL shows a pronounced asymmetry. This feature is similar to direct ¹³C-DNP using TOTAPOL and the two enhancement profiles for ²H and ¹³C DNP coincide with the maximum absolute enhancement observed on the low-field side (DNP(-)). This appears to be an inherent feature of TEMPO-based polarizing agents, when low- γ nuclei such as ¹³C and ²H are polarized. This is in contrast to ¹H-DNP where the maximum absolute enhancement is observed on the highfield side (DNP(+)).

For ¹H DNP, the TEMPO based biradical TOTAPOL currently yields the largest enhancements in DNP-enhanced MAS-NMR experiments.^{38,41} However, with an inhomogeneous breadth of $\Delta \approx 600$ MHz at 5 T, TOTAPOL is not optimized for polarizing low- γ nuclei such as ²H, ¹³C or ¹⁵N and polarizing agents with narrower EPR spectra are preferable. At present only two radicals are known for DNP applications that have a narrow EPR spectrum at high magnetic fields, the

stable trityl radical and its derivatives^{42,43} and 1,3-bis-(diphenylene)-2-phenylallyl (BDPA).⁴⁴ Here we choose the trityl radical OX063 (see Fig. 1) as the polarizing agent, because of its copious solubility in aqueous media.⁴⁵ The 140 GHz EPR spectrum of OX063 is shown in Fig. 2 (top). The spectrum is essentially symmetric with a spectral breadth of $\Delta \approx 55$ MHz (FWHM) as determined from the EPR spectrum. As a consequence, the enhancement profile of OX063 for direct ²H-DNP shown in Fig. 2 is symmetric with the maximum positive enhancement occurring at 4983.0 mT (DNP(+)) and the maximum negative enhancement occurring at 4980.7 mT (DNP(-)).

A direct comparison of these two enhancement profiles can be used to illustrate another important fact for high-field DNP. At 5 T the separation between the optimum field positions for ¹H-DNP using TOTAPOL (DNP(+)) and ²H-DNP (or ¹³C) is approximately 4 mT, corresponding to \sim 112 MHz electron Larmor frequency. The separation is 14 mT between DNP(-) for TOTAPOL and DNP(+) for OX063, corresponding to 400 MHz for electrons. To be able to study different polarizing agents and to cover the complete field range, the DNP spectrometer has to be equipped either with a sweep coil or the gyrotron needs to be tunable over a range of > 0.5 GHz.^{46–48} Note that the sweep/tuning range will increase at higher fields.

A comparison of the ²H-DNP performance for TOTAPOL and OX063 is shown in Fig. 3 and approximately a factor of 4 larger enhancement is observed for OX063 under similar experimental conditions. This improvement is due to the much narrower EPR spectrum of OX063 (Δ (TOTAPOL)/ Δ (OX063) \approx 11) allowing a larger fraction of the electron spins to be excited by the microwave radiation. Note that at the same electron concentration TEMPO-based biradicals give a factor of 4 larger enhancements compared to monomeric TEMPO,²³ and we therefore expect that further improvements could be made using biradicals based on OX063. Due to the much better performance of OX063 over TOTAPOL, the following DNP experiments were all performed using OX063 as the polarizing agent.



Fig. 3 Comparison of the steady-state ²H signal intensity for TOTAPOL (A) and OX063 (B). Both spectra were recorded backto-back under identical experimental conditions. Due to the insufficient excitation bandwidth of 83 kHz, the magnitude spectrum is shown. T = 90 K, $\omega_r/2\pi = 5.882$ kHz. The spectra were recorded using a rotor-synchronized quadrupole echo sequence.

Bulk-polarization build-up and maximum enhancement

During the DNP process, the high thermal electron polarization is transferred to the surrounding nuclei resulting in a bulk-polarization build-up curve that can be modeled by an exponential process with a characteristic bulk-polarization build-up time constant $\tau_{\rm B}$. Fig. 4 illustrates a ¹³C-detected bulk-polarization build-up curve for ²H DNP using OX063 as the polarizing agent. Here the DNP-enhanced ²H polarization is transferred to the proline ¹³C nuclei for detection via a subsequent cross-polarization (CP) step.49 This allows an accurate determination of the signal enhancement, because the ¹³C spectrum is much narrower compared to the direct-detected ²H spectrum. At a temperature of 90 K, the steady-state polarization is reached after approximately 100 s of microwave irradiation yielding a build-up time constant of $\tau_{\rm B}(^{2}{\rm H}) = 21 {\rm s}.$

The absolute enhancement is calculated from the microwave on and off spectra, recorded under identical experimental conditions (see Fig. 4, inset). For the off signal, 40 times more scans were averaged to provide sufficient signal-to-noise due to the small ²H signal intensity without DNP enhancement and a steady-state ²H DNP enhancement of $\varepsilon \geq 700$ was observed. Theoretically, the maximum enhancement that can be achieved in a DNP experiment is given by the ratio of the gyromagnetic ratios of the electron and the nucleus that is polarized, here ²H ($\gamma(e^{-})/\gamma(^{2}H)$). This gives a theoretical maximum enhancement of 4300 for ²H-DNP.

In Fig. 5 two direct ¹³C-DNP enhanced spectra of proline are shown, one spectrum taken without decoupling (A) and one with 83 kHz high-power ²H TPPM decoupling (B).⁵⁰ As expected, no significant difference in resolution was detected between the two acquisition schemes. Therefore, the following



Fig. 4 ²H bulk-polarization build-up curve recorded at a magnetic field position corresponding to DNP(+) using OX063. The ²H polarization is detected indirectly from the total ¹³C signal of $U-[^{2}H_{7}, {}^{13}C_{5}]$ -proline through a ramped cross-polarization step (1.5 ms), 16 scans averaged. The inset shows the mw-on and off signal. The DNP enhanced spectrum was recorded at a field position corresponding to DNP(+) with a DNP build-up time of $t_{\rm mw} = 20$ s. For the mw-on signal 32 transients were averaged while for the mw-off signal in total 1280 transients were averaged. T = 90 K, $\omega_r/2\pi = 5.882$ kHz. Spinning side bands are marked by asterisks.



Fig. 5 Direct ¹³C DNP-enhanced MAS-NMR spectra of U-[²H₇, ¹³C₅]-proline taken at 90 K, $\omega_R/2\pi = 5.5$ kHz, 4 scans. A: Spectrum taken without decoupling. B: Spectrum taken with 83 kHz of TPPM ²H decoupling.

experiments were all performed without decoupling of the (residual) 1 H or 2 H nuclei.

²H-DNP Enhanced ²H-¹³C correlation spectroscopy

Depending on the experimental conditions the electron polarization can be either used to polarize ¹³C nuclei directly ($e^- \rightarrow {}^{13}C$) or indirectly ($e^- \rightarrow {}^{2}H \rightarrow {}^{13}C$). In the second case the electron polarization is first transferred to the ²H nuclei *via* DNP and then transferred to the ¹³C nuclei by a subsequent CP step.⁵¹ In Fig. 6 two ²H-DNP-enhanced ¹³C detected MAS-NMR spectra of U-[${}^{2}H_{7}$, ${}^{13}C_{5}$]-proline recorded at 90 K are shown. The top spectrum in Fig. 6 is a direct ¹³C-DNP enhanced spectrum of proline and all five proline ¹³C resonances are visible. The second spectrum shown in Fig. 6 (bottom) is an indirect polarized ¹³C spectrum of proline.

Due to the short contact time of the CP process (1.5 ms), predominantly one-bond polarization transfer from ²H to ¹³C is observed. The ¹³C signal intensity for the carbonyl atom is attenuated due to the lack of a directly bonded deuterium, whereas nuclei that do posses a directly bonded deuterium (α - γ) yield intense lines.

The pulse sequence used for DNP-enhanced ²H doublequantum (DQ) filtered ¹³C correlation spectroscopy is shown in Fig. 7. Double quantum coherences are excited using a two-pulse scheme, ⁵² consisting of a DQ excitation and reconversion period (characterized by τ) separated by a rotor-synchronized t_1 evolution period given by $n^*\tau_R$ with *n* the number of rotor cycles and τ_R the rotor period. Finally the ²H magnetization is transferred to ¹³C by a CP step.⁵¹ For ²H-DNP-enhanced measurements, the sample is irradiated by continuous wave (CW) microwave radiation, on-resonant with the DNP transition.

The DQ efficiency is determined by comparing the signal intensity obtained from a ¹³C CPMAS experiment with the signal intensity obtained from a ²H double-quantum filtered ¹³C CPMAS experiment as shown in Fig. 8. From this comparison a ²H double-quantum efficiency of ~50% is observed.

A two-dimensional ²H-DNP enhanced ²H, ¹³C correlation spectrum of proline is shown in Fig. 9. Here a ²H



Fig. 6 ¹³C MAS-NMR spectra of U-[²H₇, ¹³C₅]-proline taken at 90 K. A: Direct ¹³C DNP-enhanced MAS-NMR spectrum, $\omega_R/2\pi = 5.5$ kHz, 4 scans, $t_{mw} = 60$ s. B: ²H DNP-enhanced ¹³C MAS-NMR spectrum. The polarization is transferred from ²H to ¹³C by a cross-polarization step (1.5 ms), $\omega_R/2\pi = 5.0$ kHz, 64 scans, $t_{mw} = 20$ s. Spinning side bands are marked by asterisks. The sensitivity of the two spectra are 7.9 and 1.3 S/N seconds^{-1/2} for A and B, respectively. The main source of sensitivity difference is due to inefficiency in the CP step in which the ²H spin lock of ~83 kHz covers less than half of the ~200 kHz broad ²H spectrum.



Fig. 7 Pulse sequence to record a ²H double-quantum, ¹³C correlation spectrum. Double quantum coherences are generated using a two-pulse sequence. The t_1 evolution time is rotor-synchronized.

double-quantum filter is used before the polarization is transferred to ¹³C through a 1.5 ms CP step. A double-quantum excitation and reconversion time of 1 μ s was used, followed by a *z*-filter of 3 μ s length. The pulse sequence used here is similar to the one previously reported for DQ-filtered ²H,¹³C correlation spectroscopy in perdeuterated proteins.¹¹ The two-dimensional



Fig. 8 Determination of the DQ efficiency for U- $[{}^{2}H_{7}, {}^{13}C_{5}]$ -proline from DNP-enhanced spectra. Top: ${}^{13}C$ CPMAS spectrum. Bottom: ²H double-quantum filtered ${}^{13}C$ CPMAS spectrum with $t_{1} = 0$. Experimental conditions: T = 90 K, $\tau = 1$ µs, $\Delta = 3$ µs, $\omega_{R}/2\pi = 5.882$ kHz, $t_{mw} = 20$ s.



Fig. 9 Two-dimensional DNP-enhanced ²H-DQ-¹³C correlation spectrum of U-[²H₇, ¹³C₅]-proline recorded at 90 K, $\omega_R/2\pi =$ 5.882 kHz, sampling time in the indirect dimension $\Delta t_1 =$ 170 µs, DQ excitation and reconversion time $\tau = 1$ µs, $\Delta = 3$ µs, $t_{mw} = 25$ s, 64 scans per t_1 point, ~10 h of total acquisition time.

spectrum shows four resolved cross-peaks, corresponding to correlations between the ¹³C proline atoms and the covalently attached ²H nuclei.

Spectral linewidths

Under the current experimental conditions linewidths of approximately 10 ppm and 8 ppm were observed for ²H and ¹³C, respectively. These linewidths are larger than those observed previously for perdeuterated proteins.^{8,11} However, the source of the increased linewidth is not of a general nature. In particular the main contribution arises from the fact that proline is a small molecule embedded in a frozen (90 K) glassy solvent matrix (glycerol/water). DNP samples are typically prepared in a glass-forming solvent, which serves as a cryoprotectant to ensure that the polarizing agent is homogeneously dispersed throughout the sample and protects proteins from cold degradation caused by thermal cycling of the sample. This is known to induce conformational distributions, which in turn can cause inhomogeneous broadening.53 However, this factor becomes unimportant for larger systems such as biomacromolecules or (nano) crystals. For example in contributions by Barnes et al. and Debelouchina et al. (same issue) ¹³C linewidths of 1–2 ppm are observed for the membrane protein bacteriorhodopsin (bR) and GNNQQNY nanocrystals.

The paramagnetic polarizing agent has only minor effects on the linewidth in such systems. For example in DNP-enhanced MAS-NMR experiments on amyloid nanocrystals GNNQQNY54 or the membrane protein bacteriorhodopsin (bR) the radical does not penetrate into the protein or nanocrystals. In the case of bR even electron concentrations of up to 100 mM did not show any effect on the linewidth of the retinal, which is buried inside the protein.⁵⁵ All experiments described here are performed at a magnetic field strength of 5 T (212 MHz for ¹H), which is rather low for contemporary MAS-NMR spectroscopy and second order quadrupole effects could have a contribution to the observed linewidth. Another minor contribution to the linewidth is of a technical nature. It is rather difficult to accurately set the magic angle at cryogenic temperatures for this particular probe, since it is not equipped with a cryogenic sample-eject system¹⁶ or a Hall effect sensor.⁵⁶ Although a misadjusted magic angle has only minor effects on the linewidth for double-quantum filtered ²H experiments^{11,14}

it nevertheless adds a contribution to the line broadening. There is also the possibility that small inhomogeneities in the magnetic field at the sample caused additional line broadening.

Sensitivity gain through DNP

Acquisition of a ²H dimension offers several advantages over a ¹H dimension. The deuterium spin system has a lower gyromagnetic ratio, and therefore does not suffer from the homogenous broadening observed for high concentrations of protons in solids. Spins of interest can be perdeuterated without deuteration of solvents, crystallization agents and cofactors. Comparable sensitivity should also be achievable with deuterium detection. For example, methyl-methyl contacts are often important for determination of protein structure, and in cases where a CD₂H labeling is used to reduce proton couplings, perdeuteration ($\sim 97\%$) is employed.⁸ At 3% protonation, methyl groups are \sim 9% CD₂H spin systems to first order, with minimal ($\sim 0.3\%$) CDH₂ and CH₃ labeling. This avoids broadening in the ¹³C dimension due to the shift in the isotropic resonance between CH and CD which results in different isotropic shifts for CH₃, CH₂D, CHD₂ and CD₃ groups. Since 10% labeling is often found to be necessary for optimized relaxation characteristics of amide protons,⁸ perdeuteration will be used as a point of comparison, but may need to be adjusted to by a factor of ~ 3 if higher protonation is found to be optimal.

In a perdeuterated sample (~97%) ²H NMR should have a factor of ~8.6 higher sensitivity compared to ¹H detection, and a factor of ~2.5 was experimentally observed by Agarwal *et al.*¹¹ If CH or CH₂ groups are of primary interest, or if a higher proton concentration is found to be optimal, this analysis needs to be adjusted. This gain in sensitivity is mainly due to the short longitudinal relaxation of the ²H nuclei, a direct consequence of the large quadrupolar coupling. Therefore, at room temperature the recycle delay in the NMR experiment can be short. Furthermore, sample heating is not an issue due to the much lower decoupling power needed for deuterium. Nevertheless recycle delays between 1.25 and 3 s were reported for previous work on biological samples.¹¹

This advantage no longer exists at 90 K because the DNP build-up time constant is 21 s. Therefore, to run the DNP experiment at the optimum repetition rate one needs to wait 21*1.25 s = 26 s between shots and the sensitivity for low-temperature ²H MAS-NMR spectroscopy would be decreased by a factor of 3 to 5 depending on the actual recycle delay used in the experiment compared to experiments performed at 300 K. However, the observed DNP signal enhancement of $\varepsilon \ge 700$ leads to an overall sensitivity of a low-temperature ²H-DNP enhanced MAS-NMR experiment that is a factor of 140 to 240 larger than at room temperature. This does not include the additional factor of ~3 in sensitivity due to the lower temperature (300 K/90 K).

To compare the overall efficiency of ²H-DNP with ¹H-DNP the degree of nuclear polarization can be compared. In the case of ²H-DNP this is 16% of the theoretical maximum, and for ¹H-DNP typically 27% (175/660) is observed at a magnetic field of 5 T.^{23,38} Therefore, overall ¹H-DNP currently performs more efficiently than ²H-DNP. However, with further advances

in polarizing agents, and despite a bulk-polarization buildup time of ~5 s for ¹H-DNP⁴⁰ and 21 s for ²H DNP, we expect both methods to be competitive on a sensitivity basis. Importantly, ²H MAS-NMR provides a facile approach to introduce a pseudo-¹H dimension into the spectra. Note that this comparison does not include the efficiency of the CP transfer.

Materials and methods

Sample preparation

Field-swept DNP enhancement profiles are recorded using a solution of 20 mM TOTAPOL or 40 mM OX063 in d_8 -glycerol/D₂O (60/40). Direct ²H signal detection was performed using a rotor-synchronized quadrupole-echo sequence.

For DNP experiments on proline, a 1.25 M solution of U-[${}^{13}C_5$, D₇]-proline in d₈-glycerol/D₂O (60/40) was prepared with 40 mM OX063 as the polarizing agent. Note that the high proline concentration is only necessary for recording the off-signal (no mw) in a reasonable amount of time. Isotopically labeled proline (U- ${}^{13}C_5$, 97–99%; U-D₇, 97–99%; ${}^{15}N$, 97–99%) was purchased from Cambridge Isotope Laboratories (Andover MA, USA). All solvent mixtures are given in weight ratios.

DNP spectroscopy

All DNP experiments were performed on a custom-designed DNP NMR spectrometer operating at a magnetic field of 5 T corresponding to a Larmor frequency of 211 MHz (¹H) and 140 GHz (e⁻), respectively. A custom-designed cryogenic MAS-NMR probe was used for radio-frequency (rf) irradiation (¹³C and ²H) with a commercial 2.5 mm spinning module (Revolution NMR Inc.). Typically a 50 kHz rf field-strength was obtained on the ¹³C channel, while the ²H field-strength was 83 kHz. ²H⁻¹³C cross-polarization was performed using a 50 kHz field on both channels for a duration of 1500 µs. All spectra are recorded without high-power ¹H or ²H decoupling (see Fig. 5).

High-power microwave radiation was generated using a gyrotron oscillator operating at 139.662 GHz,^{57,58} capable of producing high-power (>10 W) millimetre waves. The DNP sample (~6 μ L) was placed in a 2.5 mm sapphire rotor and a microwave power of 2.5 W was estimated at the position of the sample. The 5 T superconducting magnet is equipped with a superconducting sweep coil to sweep the magnetic field over a range of 750 G. For accurate field measurements, the spectrometer is equipped with a field mapping unit.

EPR spectroscopy

EPR experiments were performed on a previously described custom-designed high-field EPR spectrometer operating at a microwave frequency of 139.504 GHz.^{59,60} The sample (~250 nL, 1 mM) was placed in a Suprasil quartz tube with an outer diameter of 0.55 mm. EPR spectra were recorded with a two-pulse echo sequence ($\pi/2-\tau-\pi-\tau$ -echo) by integrating the echo intensity while sweeping the magnetic field ($t_p(\pi/2) = 60$ ns, $\tau = 300$ ns). For accurate field measurements, the spectrometer is equipped with a field/frequency lock system.⁶¹

Conclusions

We have demonstrated the application of direct ²H-DNP to two-dimensional ²H, ¹³C MAS-NMR correlation spectroscopy. A steady-state signal enhancement of $\varepsilon = 700$ was observed with a bulk-polarization build-up constant of $\tau_{\rm B} = 21$ s. Under these conditions the sensitivity of a ²H MAS-NMR experiment can be increased by two orders of magnitude, compared to ²H experiments performed at room temperature. We believe that the combination of perdeuteration and ²H-DNP could have a large impact on protein assignment and structure determination, as the deuteron can be used as an additional nucleus to introduce additional resolution and structural information about the system under study into the spectrum. We believe this approach may be widely applicable, requiring little optimization of isotopic labeling strategies. Furthermore, we expect that technical improvements in hardware and sample preparation for low-temperature MAS-NMR spectroscopy can be expected to vastly improve linewidths in future biological applications. We are currently exploring these improvements.

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