Supporting Information for

Equilibration of tyrosyl radicals $(Y_{356}, Y_{731}, Y_{730})$ in the radical propagation pathway of the *E. coli* class Ia ribonucleotide reductase

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Supporting Methods:

Ouantitation of the amount of [\beta^2H_2]Y incorporated in \alpha 2 by LC-MS. Purified $[\beta^2H_2]Y-\alpha 2$ (2) mg/ml) in 8 M urea (Sigma Aldrich), 50 mM ammonium bicarbonate (pH 7.8) was incubated with 10 mM DTT at 60 °C for 45 min. Cysteine residues of this protein were then alkylated by 22 mM iodoacetamide at room temperature (RT) in the dark for 1 h. Urea and excess reagent were removed by repeated concentration/dilution with 50 mM ammonium bicarbonate using a Microcon (Millipore) YM-10 centrifugal filtration device. Trypsin (Porcine pancreas, Sigma) was then added at an enzyme-to-substrate ratio (w/w) of 1:30, and incubated for 16 h at RT. The reaction was stopped by adjusting the pH to ~3 using 1% formic acid and the solution filtered through the Microcon YM-10 centrifugal filter to remove trypsin and any remaining undigested protein from the small peptides. The peptides were then analyzed by a nano flow high-performance liquid chromatography (HPLC) system (Eksigent Tempo) connected to an electrospray ionization mass spectometer, QSTAR Elite quadrupole time-of-flight mass spectrometer (AB Sciex). A reversed phase C₁₈ capillary HPLC column (250 x 0.075 mm, Higgins Analytical) was used. The mass spectrometer was calibrated with the fragment ions of peptide Glu-fibrinopeptide B with a mass accuracy of < 25 ppm.



Figure S1. LC-MS analysis of tryptic peptides of non-labeled wt- $\alpha 2$ and $[\beta - {}^{2}H_{2}]Y-\alpha 2$. MS of peptide fragments containing Y_{730} - and $Y_{731}-\alpha 2$, TLYYQNTR, are shown for non-labeled wt- $\alpha 2$ (A) and $[\beta - {}^{2}H_{2}]Y-\alpha 2$ (B). The fragment was detected as the doubly charged ion with m/z of 529.76 ± 0.03 for non-labeled wt- $\alpha 2$ (A), which agrees with the calculated m/z of 529.77. In the $[\beta - {}^{2}H_{2}]Y-\alpha 2$ sample, a doubly charged ion at m/z 531.77 ± 0.05 was observed with the same retention time (B). This result is consistent with the calculated m/z of 531.77 for the peptide fragment with two $[\beta - {}^{2}H_{2}]Ys$. In B, a small signal with m/z 529.76 ± 0.03 that corresponds to the non-labeled peptide, and no signal that corresponds to the peptide with one $[\beta - {}^{2}H_{2}]Y$ was observed. Based on the relative intensities of these peaks, the $[\beta - {}^{2}H_{2}]Y$ incorporation was estimated to be > 92%. The sequence of peptides with m/z 529.76 observed in (A) and the peptide with m/z 531.77 in (B) were further analyzed by fragment ion MS to confirm that the observed signals are associated with the expected TLYQQNTR peptide.



Figure S2. Effect of freezing method (A, B) and protein concentration (C, D) on the PELDOR spectra. (A) PELDOR spectra of 0.2 mM [NO₂Y₁₂₂•]- β 2 prepared by hand-quenching with 5% glycerol (red) or 30% glycerol (yellow) and by rapid freeze quenching with 5% glycerol (blue), and with 0.2 mM wt- β 2 prepared by rapid freeze quenching with 5% glycerol (green). All traces are normalized by the signal intensity at t = 0; (B) Spectra corrected by subtraction of monoexponential signal decay function and fit by Tikhonov regularization procedure¹; (C) PELDOR spectra of rapid freeze quenched samples prepared with 0.2 (red), 0.1 (black) and 0.05 mM (blue) [NO₂Y₁₂₂•]- β 2 and α 2/CDP/ATP and quenched at 24 ms. All traces are normalized by the signal intensity at t = 0; (D) The analysis is as described in B.



Figure S3. 9 GHz CW-EPR spectra of the reactions with $[NO_2Y_{122}\bullet]$ - $\beta 2$ (0.1 mM), wt- $\alpha 2$ (0.1 mM), ATP (3 mM) and CDP (1 mM) at 25 °C and freeze quenched at (A) 8 or (B) 24 ms. The subtraction of NO₂Y• spectrum (green, 79 and 66% of total radical in (A) and (B), respectively) from the observed spectrum (blue) resulted in pathway Y• spectrum (red, 21 and 34 % in (A) and (B), respectively). Note that the NO₂Y• feature is broader than the Y• feature on the low field side facilitating subtractions.



Figure S4. Time course of NO₂Y• reduction and Y• formation with $[\beta^{-1}H_2]Y-\alpha^2$ (blue, NO₂Y• and orange, Y•) and $[\beta^{-2}H_2]Y-\alpha^2$ (green, NO₂Y• and red, Y•) at 25 °C monitored by RFQ-EPR spectroscopy. Sequential mixing RFQ was carried out as described in Methods². Each point represents an average of three replicates.



Figure S5. (A) Numbering for atoms of tyrosine. The p_z orbital on C1 is indicated. The amino and carboxy groups attached to C α are omitted for clarity. (B) Definition of the ring rotation angle θ in tyrosine. θ (-90° ~ +90°) represents the dihedral angle between C_{β}-H_{pro-S} bond and the p_z orbital on the C1 atom of the phenyl ring of Y.



Figure S6. Quantitation of $[\beta^{-2}H_2]Y \cdot$ at 12, 24 and 60 ms at 25 °C. EPR spectra of the pathway $Y \cdot$ in the $[NO_2Y_{122} \cdot] -\beta 2/[\beta^{-2}H_2]Y \cdot \alpha 2/ATP/CDP$ reaction (red traces) were reconstructed using $Y_{356} \cdot$ observed in the $[NO_2Y_{122} \cdot] -\beta 2/[Y_{731}F] \cdot \alpha 2/ATP/CDP$ reaction (blue) and that of $[\beta^{-2}H_2]Y \cdot$ simulated using the parameters shown in Table 1 (pink). The sum of the $Y_{356} \cdot$ and $[\beta^{-2}H_2]Y \cdot$ spectra is shown in the black traces. To determine the amount of $[\beta^{-2}H_2]Y \cdot$, the ratio of $Y_{356} \cdot$ and $[\beta^{-2}H_2]Y \cdot$ spectra was adjusted to minimize the difference between the red and the black traces. The ratios of $Y_{356} \cdot$ and $[\beta^{-2}H_2]Y \cdot$ determined by this analysis are indicated. Other details of the analyses are described in the main text.



Figure S7. EPR spectral simulation for $[\beta^{-2}H_2]Y$ • with hfcs for $\beta^{-2}H$ of 6 (blue), 8 (red), and 10 MHz (green) with additional parameters summarized in Table 1 in the main text.



Figure S8. An Arrhenius plot of the NO₂Y• reduction in the reaction of $[NO_2Y_{122}\bullet]-\beta2/wt-\alpha2/ATP/CDP$ monitored by SF-absorption spectroscopy at 460 nm as described previously². Rate constants were determined from 4-7 traces. Briefly, Fe(II)₂- $[NO_2Y_{122}]-\beta2$ (40 µM, 5 Fe(II)/ $\beta2$) in anaerobic 50 mM HEPES (pH 7.6) in syringe A was mixed with an equal volume of O₂-saturated 50 mM HEPES (pH 7.6) containing CDP (4 mM) in syringe B and aged for 0.5 - 2 s to maximize NO₂Y• formation in the incubation loop. The resulting solution was then mixed with equal volumes of wt- $\alpha2$ (20 µM), ATP (6 mM), EDTA (2 mM) and MgSO₄ (30 mM) in 50 mM HEPES from syringe C. The final reaction mixture contained 10 µM [NO₂Y₁₂₂]- $\beta2$ and $\alpha2$, 3 mM ATP, 1 mM CDP, 1 mM EDTA and 15 mM MgSO₄ in 50 mM HEPES. Experiments were carried out at 5, 10, 15, 20, 25 and 37 °C. In all cases, the kinetics were biphasic except for 37 °C, where the fast phase occurred within the mixing dead time. Lines are the linear least squares fit to log (k) = log (A) – E_a / (R x T), where k is a rate constant, E_a is the activation energy, R is the gas constant, and T is the temperature in K.

Table S1. Dihedral angles (θ) for Ys in $\alpha 2$. ^{<i>a</i>}			
	Y ₇₃₀ ^b	Y ₇₃₁ ^b	PDB ID
<i>E. coli</i> wt	-67 °	+3 °	2X0X ³
E. coli Y ₇₃₀ NH ₂ Y mutant	-67 °	+82 °	$2XO4^4$
Yeast	-58 °	+53 °	2WGH ⁵
Human	-55 °	+61 °	$2EUD^{6}$
Salmonella typhimurium	-68 °	+19 °	$1 PEQ^7$

^{*a*} Dihedral angles (θ , see Figure S5 for definition) between the C_{β}-H bond and the p_z orbital on C1 of Y_{730/731}- α found in the crystal structure of class Ia α from *E. coli*, *S. cerevisiae* and human and class Ib α from *S. typhimurium*. ^{*b*} Numberings for *E. coli* α are used.

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